

Nuclear Energy

Nuclear Technology Research and Development



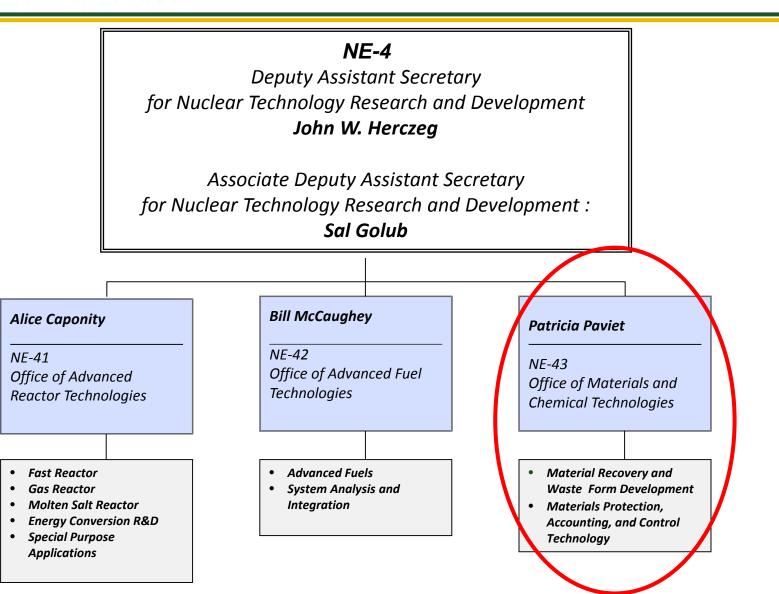
Material Recovery and Waste Form Development Campaign

Patricia Paviet, Director, Office of Materials and Chemical Technologies

NEUP Webinar August 10, 2018

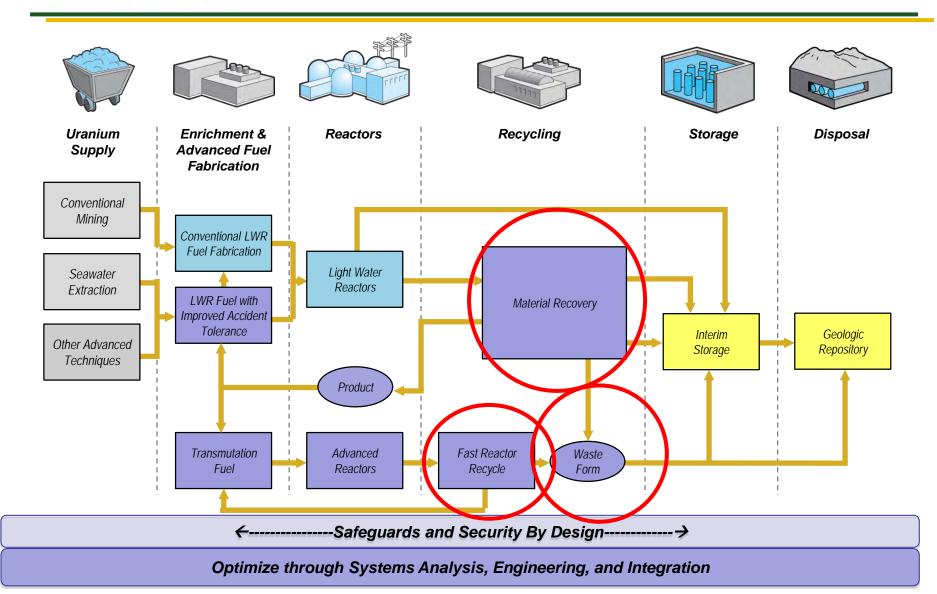


NE-4 Organization Structure





Nuclear Technology Research & Development





Material Recovery and Waste Form Development Campaign Objectives

Nuclear Energy

- Develop advanced fuel cycle separation and waste management technologies that improve current fuel cycle performance and enable a sustainable fuel cycle, with minimal processing, waste generation, and potential for material diversion
 - A key objective is to establish collaborative research programs that *maintain US fuel cycle expertise* and *provide the next generation of scientists* for nuclear science and research that are critical to implementing long term nuclear energy and waste management strategies
 - The expertise and capabilities fostered under the MRWDF campaign have been recognized throughout the US and internationally for their excellence and leadership

Maintaining Recycling Technology Expertise

Keeping the options open



Objectives of Major R&D Areas

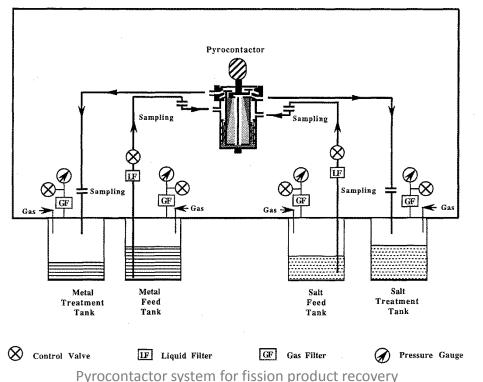
Sigma Team for Advanced Actinide Recovery (STAAR)	 Enabling technology for TRU recycle options from LWR fuel Develop cost effective technology ready for deployment
Off-gas Sigma Team	 Enabling technology for any recycle option Develop cost effective technology ready for deployment
Advanced Waste Forms and Characterization	 Enable broader range of disposal options with higher performance waste forms Develop cost effective technology ready for deployment
Electrochemical Processing	• Develop and demonstrate deployable and sustainable technology for fast reactor fuel recycling



FC-1.1: ELECTROCHEMICAL SEPARATIONS (Federal POC – Stephen Kung & Technical POC – Mark Williamson)

Nuclear Energy

- Develop innovative fission product recovery technologies to enable salt recycle thus minimizing high-level waste production and potentially reducing fuel cycle costs
 - Proposal should address the chemical basis for the recovery process, fission product elements targeted by the process, expected recovery efficiencies, final form of fission product elements for encapsulation in waste forms, and waste generation estimates



Processes could employ electrochemical, reductive extraction or other techniques to recover fission product elements, present as chlorides in the electrolyte salt, in a form suitable for encapsulation in robust waste forms.

6



FC-1.2: MATERIALS RECOVERY (Federal POC – Jim Bresee & Technical

Nuclear Energy

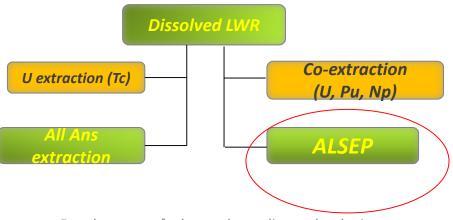
POC – Terry Todd)

Solvent Extraction Chemistry and Radiation Chemistry

Investigation of the stability of the Actinides/Lanthanides Separation Process (ALSEP) and other actinide solvent extraction systems to chemical and radiolytic degradation.

Information is needed on degradation pathways and product species.

Concepts for solvent cleanup should be developed based on the resulting insights.



Development of advanced recycling technologies



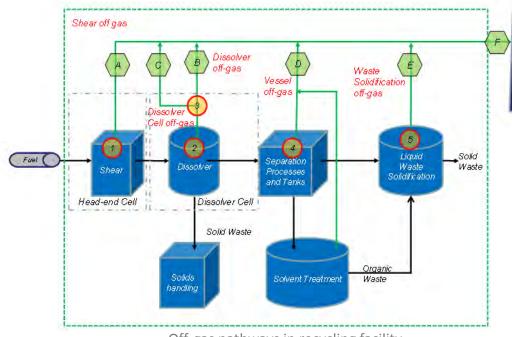
FC-1.3: ADVANCED WASTE FORMS (Federal POC – Kimberly Gray & Technical POC – John Vienna)

Nuclear Energy

Iodine Capture from Vessel Off-Gas Streams

The capture of iodine from vessel off-gas streams (VOG) is a high priority research area. It is estimated that only 1 to 6 % of the total iodine is found in this stream. However, capture of 99.9+% of this iodine is required to achieve the overall plant iodine abatement requirements. This capture is complicated by three factors: 1) The iodine concentration is 100 to 1000 times more dilute than in the dissolver off-gas steam (DOG), resulting in VOG iodine concentrations between 5 and 100 ppb. 2) The VOG gas stream is ~10 times the volume of the DOG resulting in the need for larger equipment. 3) The primary form of

the iodine in the VOG is a mixture of organic iodine species



Proposals are sought to determine the reaction pathways and kinetics for the adsorption of iodine on a silvercontaining sorbent over the range of anticipated organic iodide compounds (C1 [methyl-iodide] to C12 [iodododecane]).

The effects of temperature and associated VOG constituents on the reaction pathways and rates should also be determined.

Off-gas pathways in recycling facility



FC-1.3: ADVANCED WASTE FORMS (Federal POC – Kimberly Gray & Technical POC – John Vienna)

Nuclear Energy

Zeolite Formation Thermodynamics and Kinetics

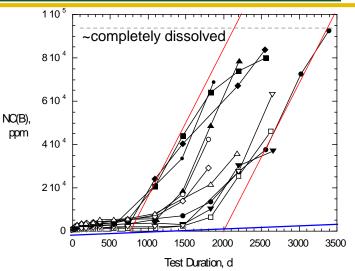
The nucleation and growth of zeolite secondary phases during borosilicate waste glass degradation is believed to couple with the dissolution kinetics of the glass and increase the dissolution rate under certain conditions.

Identifying solution conditions conducive to the formation of rate-affecting phases will allow the long-term behavior of borosilicate waste glasses to be modeled more accurately.

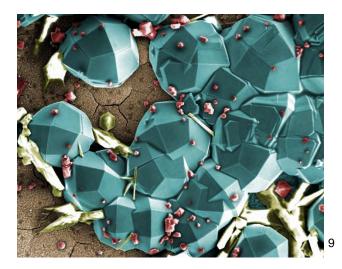
Threshold concentrations required to generate zeolites must be determined to support modeling behavior over temperature ranges of 25 to 90 °C and pH 8 to 13.

Proposals are sought to determine the composition/temperature/pH boundaries for the formation of alumino-silicate zeolites that have been identified to impact borosilicate waste glass corrosion rate and to determine the rates of precipitation as functions of the same parameters.

SEM micrograph of analcime & other alteration products on corroded glass, Jiricka et al. JNCS 2001, 292 (1-3), 25-43



Glass corrosion with acceleration, Ebert & Jerden, 2016. FCRD-MRWFD-2016-000296, ANL





Summary

- The NTR&D Programs are looking forward to partnering with universities to enhance their R&D portfolio and research capabilities
- This call is tailored to research topics that are well suited for university research
- The MRWFD program seeks university researchers who want to actively participate in the program and enhance interactions with national laboratory research staff
- The Material Recovery and Waste Form Development management team considers NEUP Principal Investigators to be an integral part of our research programs!
 - We encourage and actively seek close engagement with the campaigns





Contact Information

- Patricia Paviet: <u>patricia.paviet@nuclear.energy.gov</u>
- Jim Bresee (Aqueous Recycling): james.bresee@nuclear.energy.gov
- Kimberly Gray (Waste-Form and Off Gas): <u>kimberly.gray@nuclear.energy.gov</u>
- Stephen Kung (Electrochemical Separations): <u>stephen.kung@nuclear.energy.gov</u>
- Terry Todd: <u>terry.todd@inl.gov</u>
- John Vienna (Wasteform): john.vienna@pnnl.gov
- Mark Williamson (Electrochemical Separations): <u>williamson@anl.gov</u>
- Bob Jubin (Off Gas Capture and Immobilization): jubinrt@ornl.gov